# LJ UNIVERSITY

# LJ INSTITUTE OF PHARMACY

# **SEMESTER: II**

**Subject Name: Pharmaceutical Organic Chemistry-II** 

**Subject Code: BP203TP** 

**Scope:** This subject deals with general methods of preparation and reactions of some organic compounds. Reactivity of organic compounds are also studied here. The syllabus emphasizes on mechanisms and orientation of reactions. Chemistry of fats and oils are also included in the syllabus.

**Objectives:** Upon completion of the course student shall be able to

- 1. Write the structure, name and the type of isomerism of the organic compound
- 2. Write the reaction, name the reaction and orientation of reactions
- 3. Account for reactivity/stability of compounds,
- 4. Prepare organic compounds

## **Teaching scheme and examination scheme:**

Teaching Scheme				Evaluation Scheme			
Theory	Tutorial	Practical	Total	Theory		Practical	
Theory				External	Internal	External	Internal
3	1	4	6	75	25	35	15

### **Course Content:**

General methods of preparation and reactions of compounds superscripted with asterisk (\*) to be explained To emphasize on definition, types, classification, principles/mechanisms, applications, examples and differences

Sr. No.	Course Contents						
1.	Carbonyl compounds* (Aldehydes and ketones): Nucleophilic addition,	10					
	Electromeric effect, aldol condensation, Crossed Aldol condensation, Cannizzaro						
	reaction, Crossed Cannizzaro reaction, Benzoin condensation, Perkin condensation,						
	qualitative tests, Structure and uses of Formaldehyde,						
	Paraldehyde, Acetone, Chloral hydrate, Hexamine, Benzaldehyde, Vanilin,						
	Cinnamaldehyde						
2	Phenols*:						
	Acidity of phenols, effect of substituents on acidity, qualitative tests, Structure and						
	uses of phenol, cresols, resorcinol, naphthols						
	Carboxylic acids*: Acidity of carboxylic acids, effect of substituents on acidity,						
	inductive effect and qualitative tests for carboxylic acids, amide and ester						
	Structure and Uses of Acetic acid, Lactic acid, Tartaric acid, Citric acid, Succinic acid.						
	Oxalic acid, Salicylic acid, Benzoic acid, Benzyl benzoate, Dimethyl phthalate,						
	Methyl salicylate and Acetyl salicylic acid						
	Aromatic Acids*:						
	Acidity, effect of substituents on acidity and important reactions of benzoic acid.						
2	Amines:						
3	Aliphatic amines* - Basicity, effect of substituent on Basicity. Qualitative test,						

	Structure and uses of Ethanolamine, Ethylenediamine, Amphetamine					
	Aromatic Amines*:					
	Basicity of amines, effect of substituents on basicity, and synthetic uses of aryl					
	diazonium salts					
	Fats and Oils:					
4	• Fatty acids – reactions.					
	Hydrolysis, Hydrogenation, Saponification and Rancidity of oils, Drying oils.	05				
	Analytical constants – Acid value, Saponification value, Ester value, Iodine value,					
	Acetyl value, Reichert Meissl (RM) value – significance and principle involved in					
	their determination.					
	Polynuclear hydrocarbons:					
5	• Synthesis, reactions					
	• Structure and medicinal uses of Naphthalene, Phenanthrene, Anthracene,	08				
	Diphenylmethane, Triphenylmethane and their derivatives					
Total Hours						

# **Practical**

#### **Practical**

# **Experiments involving laboratory techniques**

- Recrystallization
- Steam distillation

## **Determination of following oil values (including standardization of reagents)**

- Acid value
- Saponification value
- Iodine value

## Preparation of compounds

- Benzanilide/Phenyl benzoate/Acetanilide from Aniline/Phenol/Aniline by acylation reaction.
- 2,4,6-Tribromo aniline/Para bromo acetanilide from Aniline/
- Acetanilide by halogenation (Bromination) reaction.
- 5-Nitro salicylic acid/Meta di nitro benzene from Salicylic acid / Nitro benzene by nitration reaction
- Benzoic acid from Benzyl chloride by oxidation reaction.
- Benzoic acid/ Salicylic acid from alkyl benzoate/ alkyl salicylate by hydrolysis reaction.
- 1-Phenyl azo-2-napthol from Aniline by diazotization and coupling reactions.
- Benzil from Benzoin by oxidation reaction.
- Dibenzal acetone from Benzaldehyde by Claison Schmidt reaction
- Cinnammic acid from Benzaldehyde by Perkin reaction
- P-Iodo benzoic acid from P-amino benzoic acid

# **Recommended Books (Latest Editions)**

- 1. Organic Chemistry by Morrison and Boyd
- 2. Organic Chemistry by I.L. Finar, Volume-I
- 3. Textbook of Organic Chemistry by B.S. Bahl & Arun Bahl.
- 4. Organic Chemistry by P.L.Soni
- 5. Practical Organic Chemistry by Mann and Saunders.
- 6. Vogel's text book of Practical Organic Chemistry
- 7. Advanced Practical organic chemistry by N.K.Vishnoi.
- 8. Introduction to Organic Laboratory techniques by Pavia, Lampman and Kriz.